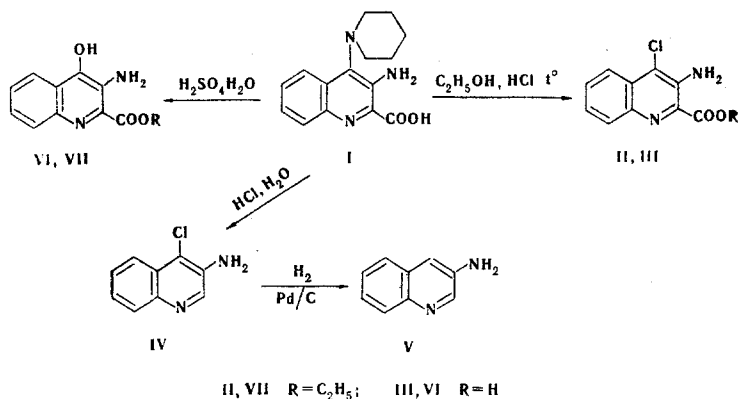


# REPLACEMENT OF A PIPERIDINE RESIDUE BY A HALOGEN IN 3-AMINO-4-PIPERIDINOQUINOLINE-2-CARBOXYLIC ACID

N. E. Britikova, L. A. Belova,  
and A. S. Elina

UDC 547.822.3'831.4.6.9

In a study of the properties of 3-amino-4-piperidinoquinoline-2-carboxylic acid (I) [1] it was found that it is possible to replace the piperidine residue in it by chlorine. Thus ethyl 3-amino-4-chloroquinoline-2-carboxylate (II) with mp 97-98° (from aqueous alcohol), is formed when dry HCl is bubbled into a refluxing solution of I in ethanol; the PMR spectrum of II (in CCl<sub>4</sub>) does not contain signals of the protons of the CH<sub>2</sub> groups of the piperidine ring but does contain multiplets of the protons of the benzene ring (7.4 and 7.8 ppm). Saponification of acid II with 3% aqueous NaOH solution gave acid III with mp 154-155° (from water). The piperidine residue is also replaced by chlorine when I is heated in dilute hydrochloric acid, but the carboxyl group is split out under these conditions to give 3-amino-4-chloroquinoline (IV) with mp 142-143° (from 3-aminoquinoline (V) with mp 84° (from benzene-petroleum ether), and by reductive dehalogenation in alcoholic alkali in the presence of Pd/C. Heating acid I in 50% sulfuric acid gave 3-amino-4-hydroxyquinoline-2-carboxylic acid (VI) with mp 267-268° (purified by reprecipitation through the Na salt), which was converted to the corresponding ester (VII), with mp 174-175° (from aqueous alcohol), by heating in ethanol with simultaneous bubbling of HCl into the solution. The experimental results make it possible to assume that protonation of I takes place at the nitrogen atom of the piperidine residue, inasmuch as the presence of a positively charged piperidinium residue in the 4 position should facilitate nucleophilic substitution at the ring C<sub>4</sub> atom to a considerable degree.



The results of elementary analysis of the compounds obtained in this study are in agreement with the calculated values. The IR spectrum of V was in agreement with the literature data.

## LITERATURE CITED

1. N. E. Britikova, L. A. Belova, O. Yu. Magidson, and A. S. Elina, *Khim. Geterotsikl. Soedin.*, 131 (1974).

All-Union Scientific-Research Pharmaceutical-Chemistry Institute, Moscow. Translated from *Khimiya Geterotsiklicheskikh Soedinenii*, No. 11, pp. 1575-1576, November, 1975. Original article submitted February 11, 1975.

©1976 Plenum Publishing Corporation, 227 West 17th Street, New York, N.Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$15.00.